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ENVIRONMENTAL
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Research Article

Comparative studies of the NO_x impacts on the photooxidation mechanisms of isomeric monoterpenes of β -pinene and limonene

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ARTICLE INFO

Article history:

Received 4 June 2024

Revised 30 July 2024

Accepted 4 August 2024

Available online 13 August 2024

Keywords:

Secondary organic aerosol

Anthropogenic–biogenic

interactions

Photooxidation mechanism

 β -pinene

Limonene

ABSTRACT

It is highly challenging to precisely compare the impacts of anthropogenic pollutants on the photooxidation of isomeric volatile organic compounds with respect to molecular compositions and particle number/mass concentrations of secondary organic aerosols (SOAs). In this study, we conducted a series of well-defined indoor chamber experiments to compare the effects of NO_x (NO and NO₂) on the photooxidation of isomeric monoterpenes of β -pinene and limonene. For the photooxidation of β -pinene with NO_x, the increase of the initial concentrations of NO ([NO]₀) shows a monotonous suppression of the particle mass concentration, whereas the increase of [NO₂]₀ shows a monotonous enhancement of the particle mass concentration. For the photooxidation of limonene with NO_x, the increase of [NO]₀ exhibits a monotonous suppression of the particle mass concentration, whereas the increase of [NO₂]₀ shows a parabolic trend of the particle mass concentration. Utilizing a newly developed vacuum ultraviolet free electron laser (VUV-FEL), the online threshold photoionization mass spectrometry reveals a series of novel compounds at molecular weight (MW) = 232 and 306 for the β -pinene + NO_x system and MW = 187, 261, 280, and 306 for the limonene + NO_x system. The molecular structures and formation pathways of these species were inferred, which led to the prediction of the diversity and difference of SOA products (i.e., ester and peroxide accretion products) formed from different monoterpene precursors.

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To improve the predictions of future air quality, it is recommended that climate models should incorporate the NO_x-driven diurnal photooxidation of monoterpenes for SOA formation mechanisms.

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Introduction

Secondary organic aerosols (SOA), primarily derived from the oxidative processes of biogenic precursors, constitute a significant fraction of the atmospheric aerosols in the troposphere, exerting substantial impacts on Earth's radiation balance, climate, and human health (Hallquist et al., 2009; Jimenez et al., 2009). Biogenic volatile organic compounds (BVOC), due to their large emissions and high reactivity, play a pivotal role in the formation of secondary pollutants (Wu et al., 2020). Laboratory and field studies indicate that the interaction between BVOCs and anthropogenic gases substantially alters the reaction characteristics of SOA formation, affording insights into the enhancement effects (Shilling et al., 2013; Xu et al., 2021, 2015; Yee et al., 2020). Understanding the interaction between BVOC and anthropogenic pollutants was proposed to be crucial for setting up effective atmospheric pollution control policies (Nagori et al., 2019).

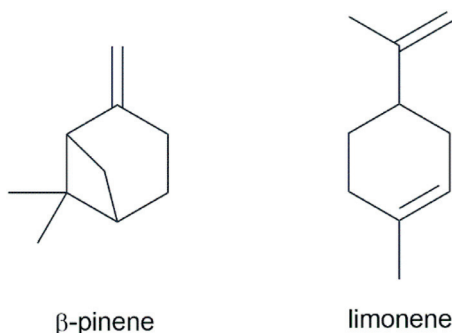
Monoterpenes, featured by high emissions (> 100 Tg C/year) and high SOA yields, are significant contributors to global biogenic SOA (Engelhart et al., 2008). Recently, more and more laboratory and field studies focus on the significance of monoterpenes in the formation of SOA (Byron et al., 2022). β -pinene and limonene, released by plants, are the primary global contributors to monoterpenes and play a crucial role as precursors in the formation of SOA (Guenther et al., 2012). As isomers, β -pinene and limonene have identical molecular formula. As shown in Scheme 1, β -pinene possesses a double-ring configuration with an exocyclic double bond, whereas limonene possesses a single-ring structure with both an external and internal double bond with enhanced reactivity as compared to β -pinene (Grosjean et al., 1993; Liu et al., 2022). Consequently, limonene exhibits higher SOA yields as compared to β -pinene (Friedman and Farmer, 2018; Liu et al., 2022). Recent experimental studies have shown pronounced differences in the formation of highly oxygenated organic molecules (HOMs) and low-volatility compounds during the

ozonolysis processes of β -pinene and limonene (Liu et al., 2022; Piletic and Kleindienst, 2022).

As an important component of anthropogenic pollutants, NO_x has been proved to exhibit a great impact on the photooxidation of BVOC (Park et al., 2017; Sarrafzadeh et al., 2016; Zhao et al., 2018). In recent years, a large number of studies indicate that NO_x suppresses the formation of new particles, thereby inhibiting the condensation of low-volatility compounds onto the surfaces (Li et al., 2017; Wildt et al., 2014). NO_x directly influences the BVOC oxidation by altering the oxidant levels. Besides, there is also indirect influence of NO_x on the chemistry of RO₂, which further affects organic compound oxidation. However, there is a lack of comprehensive comparison of the impact of anthropogenic pollutants on the photooxidation of isomeric monoterpenes of β -pinene and limonene with respect to molecular compositions and particle number/mass concentrations of SOA.

HOMs were initially identified in forests, which are composed of six or more oxygen atoms that undergo autoxidation in the gaseous phase and play a crucial role in the growth of SOA (Bianchi et al., 2019; Tröstl et al., 2016). Numerous investigations have revealed the significance of HOMs in the formation and growth of new particles (Bianchi et al., 2016; Jokinen et al., 2014; McFiggans et al., 2019). HOMs could be formed through an intramolecular hydrogen-atom shift (H-shift) within peroxy radicals and their subsequent reaction with molecular oxygen (Bianchi et al., 2019; Crounse et al., 2013). The longevity of peroxy radicals facilitates consecutive H-shift during the autoxidation processes. Considering the variation of products, lifetimes, and reactivity of HOMs in different types of SOA, it is essential to distinguish the compositional characteristics. Much efforts have been made on the formation of HOMs through the OH- and O₃-induced oxidation of monoterpenes, such as α -pinene, β -pinene, and limonene (Berndt et al., 2016; Kirkby et al., 2016; Luo et al., 2023; Shen et al., 2022; Xu et al., 2022). Given the persistently worsening NO_x pollution, it is of vital importance to explore deeply into the interrelationships between NO_x concentrations, BVOC types, and HOMs distributions, as well as the NO_x-dependent mechanisms for SOA and HOMs formation.

The objective of this study is to compare the distinct impacts of NO_x concentration on the size distribution, number concentration, mass concentration, and chemical composition of SOA formed through the photooxidation of β -pinene and limonene. Utilizing the aerosol mass spectrometry based on near-threshold photoionization using a tunable vacuum ultraviolet free electron laser (VUV-FEL), the chemical compositions of SOA were systematically analyzed. Using quantum chemical computations, we proposed plausible formation mechanisms for the newly observed products, highlighting the pivotal role of oxygen-based peroxy radical chemistry in the formation of HOMs. Our study shows that incorporat-



Scheme 1 – Structures of β -pinene and limonene.

ing the NO_x concentration-driven monoterpene SOA formation and evolution mechanisms into future climate models would lead to more accurate air quality predictions and management.

1. Materials and methods

The experiments were conducted within a 2 m³ indoor smog chamber at the Dalian Institute of Chemical Physics, Chinese Academy of Sciences (DICP-CAS) (Zang et al., 2022). A brief overview of experimental methods is provided here, with a more detailed description available in previous research (Zang et al., 2022) and in Appendix Supplementary data. In this work, β -pinene (99%, Aladdin) or limonene (99%, Aladdin) were used as primary BVOC precursors, which were introduced into the chamber through an airline. The photooxidation experiments were carried out under 40 black lamps (General Electric Company, USA). Prior to each experiment, the chamber was rigorously cleaned using an automated system, which employed a multi-step process of air condensation, freeze-drying, and catalytic adsorption to ensure zero air output with relative humidity (RH) below 3% to ensure optimal conditions. The temperature of the chamber was held constantly at approximate 24.0 °C with a thermostatically controlled air conditioner prior to the commencement of the experiment.

The concentrations of nitrogen oxides (NO, NO₂, and NO_x) were determined using a gas analyzer (Model 42i, Thermo

Fisher Scientific, UK). O₃ was formed from photochemical reactions of NO_x. The O₃ levels were measured by another gas analyzer (Model 49i, Thermo Fisher Scientific, UK). The VOCs were analyzed employing a proton-transfer reaction mass spectrometry (PTR-QMS 3500, East & West Analytical Instruments, China). Particle number concentrations, mass concentrations, and size distributions were measured using a scanning mobility particle sizer (SMPS 3938NL76; TSI Incorporated, USA). The preassigned particle density of 1.0 g/cm³ was used to convert volume concentrations to mass concentrations. The uncertainty of mass concentrations measured by SMPS was estimated to be < 2.0% by repeating the measurement of the 369 ppb limonene + 94 ppb NO₂ photooxidation reaction (Appendix A Fig. S1).

The particles generated in the DICP-CAS chamber were transported through a silicone tube to the time-of-flight aerosol mass spectrometer (TOF-AMS) using VUV-FEL for ionization. The VUV-FEL TOF-AMS employed an aerodynamic lens system to introduce aerosols from atmospheric pressure onto a copper rod mounted a high vacuum mass spectrometer (Wang et al., 2005; Zang et al., 2022). The deposited particles were vaporized by using a cartridge heater, ionized by VUV-FEL, and detected by TOF-MS in positive ion mode. Our TOF-MS was calibrated by a series of standard organic compounds (i.e., vanillin, 1-pentadecanol, n-Eicosane, etc.). The comparison of mass spectra of heater ON and OFF indicated that the ionization of gas-phase products had a negligible interference on the ionization of particles. The threshold photoionization

Table 1 – The experimental conditions of β -pinene + NO/NO₂ and limonene + NO/NO₂ photooxidation.

VOC	[VOC] ₀ (ppb)	[NO] ₀ (ppb)	[NO ₂] ₀ (ppb)	[NO _x] ₀ (ppb)	[VOC] ₀ /[NO _x] ₀	ΔM ($\mu\text{g}/\text{m}^3$)	ΔROG (ppb)	SOA Yield (%)
β -pinene	333	0	45	45	7.4	170	302	10.1
β -pinene	369	1	118	119	3.1	333	316	19.0
β -pinene	381	1	165	166	2.3	384	335	20.7
β -pinene	358	5	285	290	1.2	566	308	33.1
β -pinene	341	5	357	362	0.9	697	325	38.7
β -pinene	339	46	0	46	7.4	118	310	6.9
β -pinene	360	120	2	122	3.0	104	322	5.8
β -pinene	389	163	1	164	2.4	101	329	5.5
β -pinene	377	263	0	263	1.4	108	370	5.3
β -pinene	365	358	9	367	1.0	10	355	0.5
β -pinene	970	0	401	401	2.4	857	868	17.8
β -pinene	976	397	6	403	2.4	654	856	13.9
β -pinene	2381	8	429	437	5.4	1943	1973	17.8
limonene	360	0	52	52	7.0	1168	334	63.1
limonene	364	0	98	98	3.7	1240	343	65.2
limonene	361	0	192	192	1.9	1363	336	73.2
limonene	388	1	252	253	1.5	1521	363	75.6
limonene	347	2	367	369	0.9	1048	315	60.0
limonene	375	51	3	54	7.0	1094	334	59.1
limonene	369	94	5	99	3.7	897	325	49.8
limonene	381	185	9	194	2.0	396	365	19.6
limonene	358	246	17	263	1.4	121	332	6.6
limonene	341	347	10	357	1.0	79	319	4.5
limonene	991	0	395	395	2.5	1726	954	32.6
limonene	982	399	7	406	2.4	1689	947	32.2
limonene	2230	9	373	382	5.4	2605	1960	24.0

[x]₀ stands for initial concentration in the smog chamber and ΔM for particle mass concentration. ΔROG stands for the amount of reacted organic gas.

of molecule was achieved by using the tunable VUV-FEL to lose an electron to produce a molecular ion M^+ . The positions of mass spectral peaks detected in this work denote the molecular weight (MW) values. Our VUV-FEL TOF-AMS is able to detect the main size distributions of particles, ranging from 30 to 2500 nm and detect particle-phase products in real-time. The threshold ionization of neutral compounds was realized by taking advantage of high selectivity via the widely tunable wavelength range of VUV-FEL (50–150 nm/8.3–24.8 eV) and high detection sensitivity of trace species via the high pulse energy ($\sim 100 \mu\text{J}$) (Zang et al., 2022; Zhang et al., 2024, 2025).

In order to understand the experimental results and to study the photooxidation mechanisms of β -pinene and limonene, quantum chemical calculations were performed at the $\omega\text{B97XD}/\text{def2-TZVP}$ level of theory employing the Gaussian 16 computational chemistry software package (Frisch et al., 2016). Transition states (TSs) were subjected to a thor-

ough optimization employing the Berny algorithm, and their integrity was further corroborated by rigorous intrinsic reaction coordinate (IRC) computations. The relative energies were refined by incorporating the zero-point vibrational energy (ZPVE) corrections.

2. Results and discussion

A series of smog chamber experiments were conducted to extensively investigate the photooxidation systems of β -pinene + NO_x and limonene + NO_x (Table 1), with the primary objective of examining the specific effects of NO_x on secondary aerosol formation under various photochemical conditions. In the previous monoterpene experiments, high VOC concentrations of ~ 300 ppb (Hanson et al., 2022; Iinuma et al., 2007; Zang et al., 2024) and ~ 1000 ppb (Park et al., 2017) have

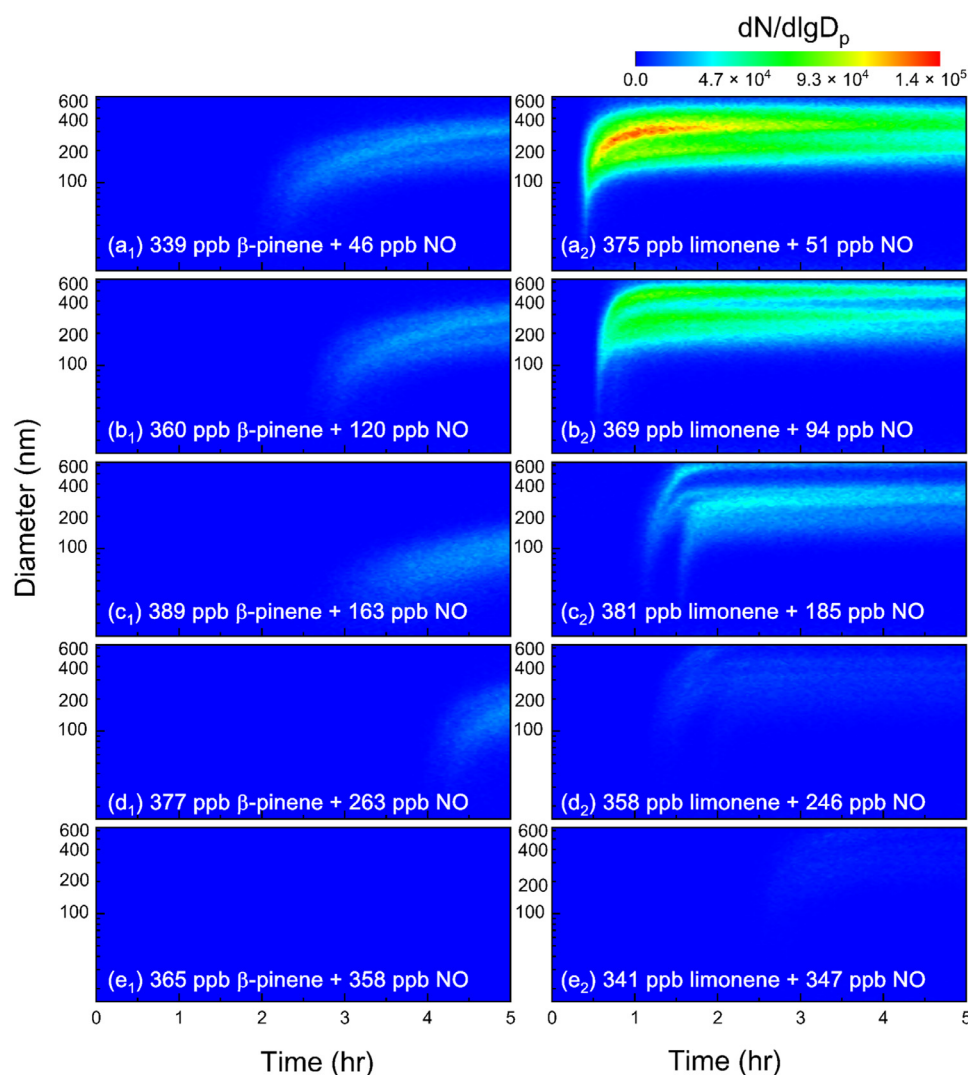


Fig. 1 – Temporal evolution of the particle size distribution in the photooxidation experiment based on $[\text{NO}]_0$ as the main variable at experimental conditions of (a₁) 339 ppb β -pinene + 46 ppb NO; (b₁) 360 ppb β -pinene + 120 ppb NO; (c₁) 389 ppb β -pinene + 163 ppb NO; (d₁) 377 ppb β -pinene + 263 ppb NO; (e₁) 365 ppb β -pinene + 358 ppb NO; (a₂) 375 ppb limonene + 51 ppb NO; (b₂) 369 ppb limonene + 94 ppb NO; (c₂) 381 ppb limonene + 185 ppb NO; (d₂) 358 ppb limonene + 246 ppb NO; (e₂) 341 ppb limonene + 347 ppb NO. D_p : particle diameter; $dN/d\lg D_p$: normalized number size distribution.

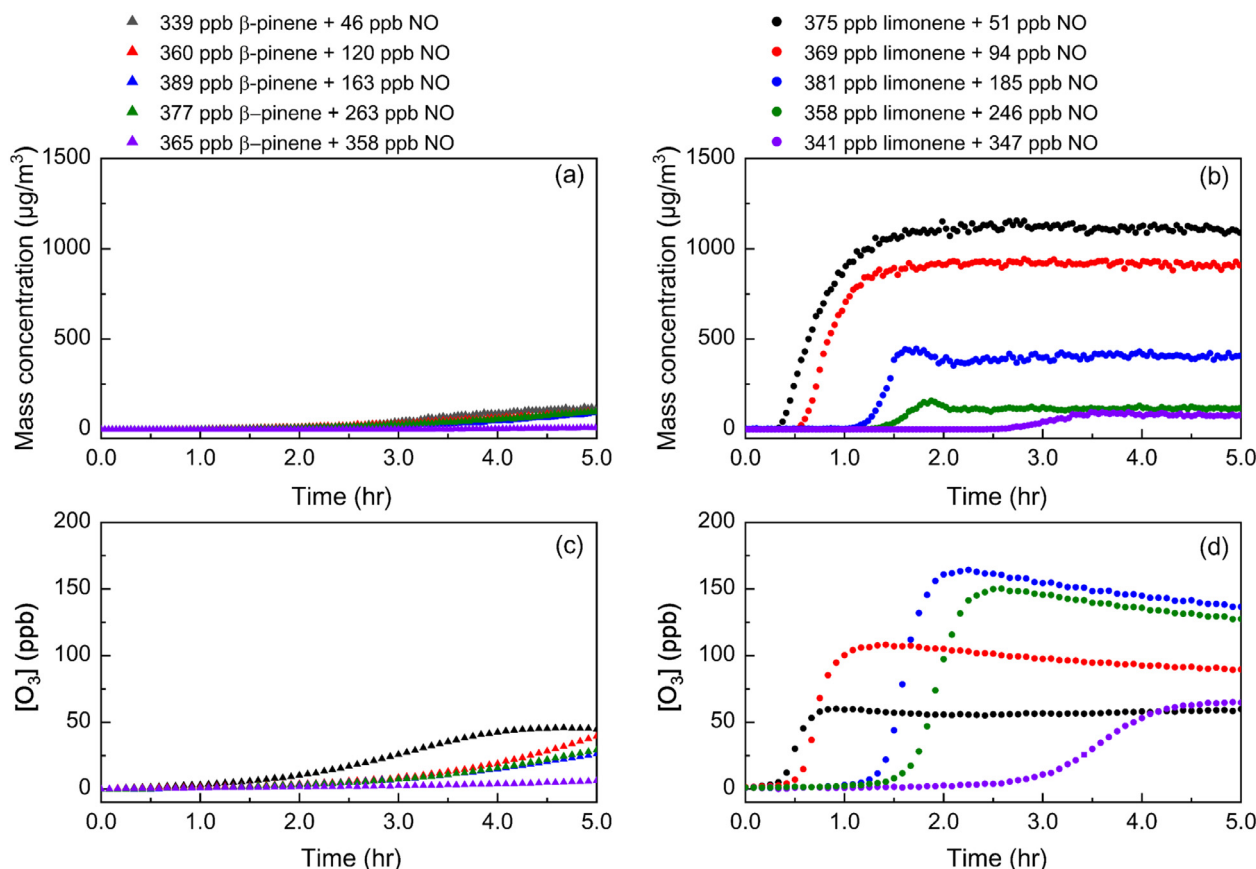


Fig. 2 – The particle mass concentration and O₃ concentration as a function of [NO]₀. The triangles and dots denote the β-pinene + NO experiments (a and c) and the limonene + NO experiments (b and d), respectively.

been used to ensure that the concentrations of particles are sufficient for accurate measurement. The isoprene concentration (100–1250 ppb) and NO_x concentration (50–600 ppb) have been set to investigate isoprene photooxidation (Zhang et al., 2011). The concentrations of 2000–4000 ppb for β-pinene and 1000–3000 ppb for limonene have also been used (Liu et al., 2022). In this work, we have selected the mixing of VOCs (300–2230 ppb) and NO_x concentration (45–406 ppb) to study the NO_x impacts on the photooxidation mechanisms of isomeric monoterpenes of β-pinene and limonene.

2.1. Effects of [NO]₀ and [NO₂]₀ on particle size distribution, number and mass concentrations

The impacts of the initial concentration of NO ([NO]₀) on the size distribution of particles formed by the β-pinene and limonene photooxidation reactions are illustrated in Fig. 1. As shown in Fig. 1a₁ and a₂, the nucleation time for the β-pinene system is 2 h in the 339 ppb β-pinene + 46 ppb NO reaction, while that for limonene is only 25 min in the 375 ppb limonene + 51 ppb NO reaction, indicating that the nucleation process in the limonene + NO photooxidation reaction is remarkably faster than that in the β-pinene + NO photooxidation reaction. Since limonene possesses a single-ring structure with both an external and internal double bond that ex-

hibits higher reactivity than β-pinene (Scheme 1), this structure may result in a faster reaction rate towards oxidation (Appendix A Table S1) (Ehn et al., 2014) and the formation of highly volatile products (Liu et al., 2022). As the [NO]₀ gradually increases, the number concentrations of SOA formed from both the β-pinene and limonene systems gradually decrease (Fig. 1a₁–e₁, 1a₂–e₂). These results indicate that NO suppresses the SOA formation, which could alter the fate of RO₂ radicals generated during VOC oxidation, lead to the formation of organic nitrates, and subsequently change the distribution of photooxidation reaction products. Such features are in line with previous studies (Chen et al., 2022; Sarrafzadeh et al., 2016; Wildt et al., 2014).

The temporal evolution of particle mass concentrations and O₃ concentrations in the β-pinene + NO and limonene + NO photooxidation experiments is illustrated in Fig. 2. The increase of [NO]₀ shows a slight suppression of particle mass concentration throughout the photooxidation reaction of β-pinene with NO (Fig. 2a). However, in the limonene + NO photooxidation experiments, as the [NO]₀ increases, there is a notably stronger suppression of particle mass concentration by influencing peroxy radical chemistry (Fig. 2b). Therefore, the dependency of SOA formation on [NO]₀ varies across different chemical systems. For photooxidation reaction of 381 ppb limonene + 185 ppb NO (Fig. 2d), the cu-

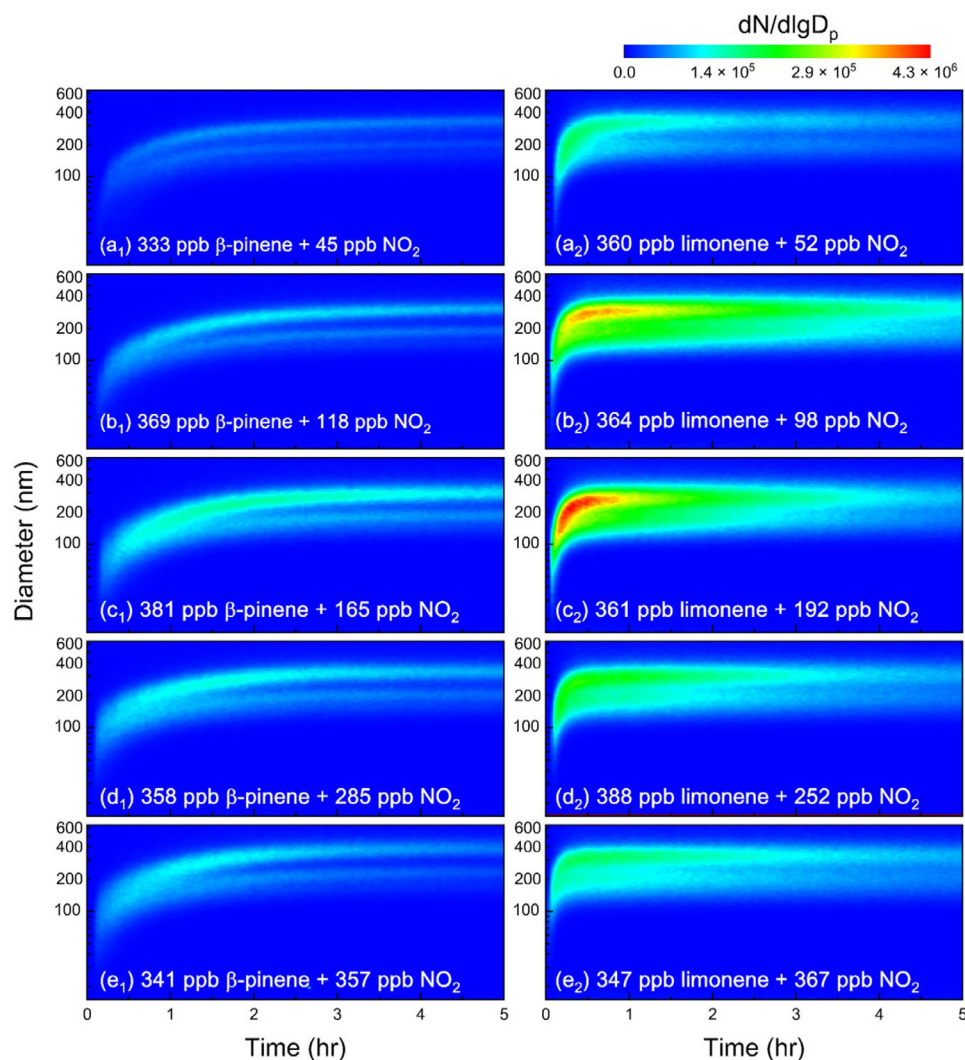


Fig. 3 – Temporal evolution of the particle size distribution in the photooxidation experiment based on $[\text{NO}_2]_0$ as the main variable at experimental conditions of (a₁) 333 ppb β -pinene + 45 ppb NO_2 ; (b₁) 369 ppb β -pinene + 118 ppb NO_2 ; (c₁) 381 ppb β -pinene + 165 ppb NO_2 ; (d₁) 358 ppb β -pinene + 285 ppb NO_2 ; (e₁) 341 ppb β -pinene + 357 ppb NO_2 ; (a₂) 360 ppb limonene + 52 ppb NO_2 ; (b₂) 364 ppb limonene + 98 ppb NO_2 ; (c₂) 361 ppb limonene + 192 ppb NO_2 ; (d₂) 388 ppb limonene + 252 ppb NO_2 ; (e₂) 347 ppb limonene + 367 ppb NO_2 . D_p : particle diameter; $dN/d\lg D_p$: normalized number size distribution.

mulative O_3 concentration is the highest, and two nucleation events are observed in Fig. 1c₂, indicative of a possible change in reaction pathway. The cumulative change in the O_3 concentration might influence the extent of the O_3 reactions with double bond of limonene, thereby affecting the types and yields of the products.

The formation of O_3 in the atmosphere is a complex chemical process (Wang et al., 2017). During the daytime, atmospheric oxidation of monoterpene is primarily driven by the interactions with NO_x , in which the NO_x photolysis plays a significant role in the generation of O_3 . The consequence of this process depends on the concentration of the oxidant and the reactivity of the monoterpene. As depicted in Fig. 2c, for the photooxidation reaction of β -pinene with NO , an increase in the $[\text{NO}]_0$ leads to a decrease in the O_3 accumulation. In the limonene + NO photooxidation reaction (Fig. 2d), the O_3

accumulation initially rises with increasing $[\text{NO}]_0$ to 185 ppb, but slows down as $[\text{NO}]_0$ levels continue to rise. These contrasting responses observed for β -pinene and limonene could be attributed to the limonene's enhanced reactivity with O_3 (Appendix A Table S1), which accelerates the production of RO_2 and subsequently enhances the conversion of NO to NO_2 , which in turn facilitates the O_3 accumulation. These findings are in line with the nonlinear variation of ground-level O_3 concentrations of NO levels (Liu and Shi, 2021).

Fig. 3 illustrates the particle size distribution of β -pinene and limonene photooxidation as a function of $[\text{NO}_2]_0$. As shown in Fig. 3a₁–e₁, in the β -pinene + NO_2 photooxidation reactions, as the $[\text{NO}_2]_0$ increases, particle number concentration gradually rises. In the limonene + NO_2 photooxidation reactions (Fig. 3a₂–e₂), as the $[\text{NO}_2]_0$ increases to 192 ppb, the particle number concentration significantly rises, but as the

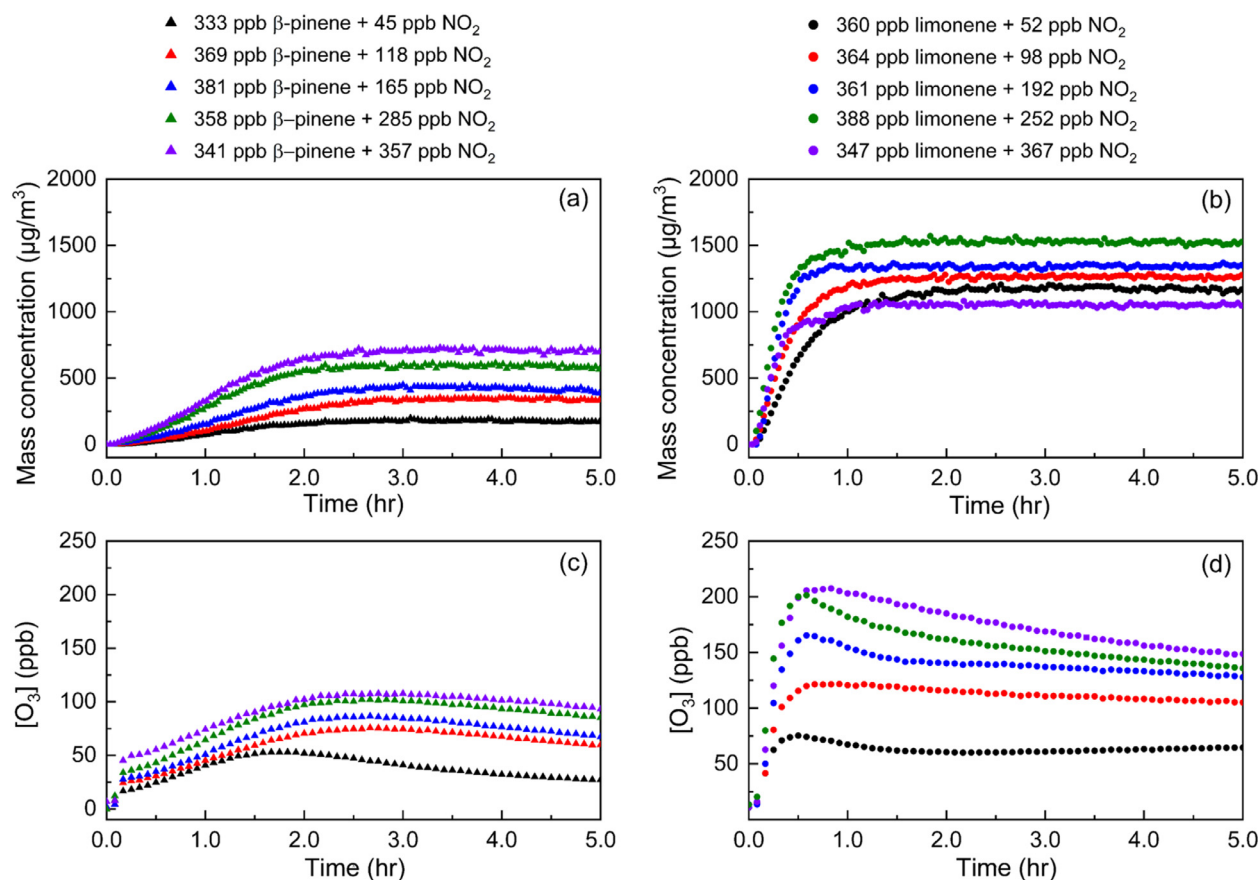


Fig. 4 – The particle mass concentration and O₃ concentration as a function of [NO₂]₀. The triangles and dots represent the β-pinene + NO₂ experiments (a and c) and the limonene + NO₂ experiments (b and d), respectively.

[NO₂]₀ continues to rise, it exhibits a pronounced suppression effect on particle number concentration. The key difference between limonene + NO₂ and β-pinene + NO₂ photooxidation reactions is ascribed to higher reactivity of limonene (preference of more RO₂ radical formation) as compared to β-pinene. The pronounced suppression effect on the photooxidation of limonene under the conditions of [NO₂]₀ > 192 ppb could be due to the reaction of RO₂ and NO produced by NO₂ photolysis. At a given [NO₂]₀, the particle size distribution of SOA formed from the limonene + NO₂ photooxidation (Fig. 3a₂–e₂) exhibits a broader range than that of SOA formed from the β-pinene + NO₂ photooxidation (Fig. 3a₁–e₁), which is reminiscent of the broader size distribution range of SOA from the limonene ozonolysis than that from the β-pinene ozonolysis (Liu et al., 2022). Such difference in the size distribution and particle number concentration for β-pinene or limonene may be related to the distribution of low volatile organic compounds, thus promoting the formation and growth of molecular clusters. Our results are supported by previously-observed high yields of extremely low-volatility organic compounds (ELVOCs) in the atmospheric oxidation of limonene (Hanson et al., 2022; Liu et al., 2022; Piletic and Kleindienst, 2022).

Photolysis of NO₂ is the main pathway for the O₃ generation in the troposphere, which directly affects the O₃ concentrations and subsequently affects atmospheric oxidation and photochemical processes (Tao et al., 2014). In the β-

pinene + NO₂ photooxidation reactions, as the [NO₂]₀ rises, the O₃ concentration generated from the NO₂ photolysis increases (Fig. 4c and Appendix A Fig. S2a₁–e₁), concurrently with an increase in the particle mass concentration (Fig. 4a). In the limonene + NO₂ photooxidation reactions, as the [NO₂]₀ increases to 252 ppb, due to the increasing accumulation of O₃ (Fig. 4d and Appendix A Fig. S2a₂–e₂), particle mass concentration also exhibits a similar increasing trend (Fig. 4b). However, as [NO₂]₀ increases to 367 ppb, particle mass concentration decreases (Fig. 4b), which is linked to the suppression effect of RO₂ formation by NO generated through the photolysis of NO₂. Since the time of O₃ cumulative peak (Fig. 4c and d) generally coincides with the time when mass concentration reaches its maximum (Fig. 4a and b), it can be inferred that O₃ triggers an increase in the mass concentration of β-pinene and limonene photooxidation reaction products. In the limonene + NO₂ photooxidation reaction, O₃ reaches its maximum at approximately 0.5 h (Fig. 4d), whereas for the β-pinene + NO₂ photooxidation reaction (Fig. 4c), O₃ reaches its maximum at approximately 2 h, which indicates that the rate of O₃ formation in the limonene photooxidation process is faster. As compared to the β-pinene + NO₂ photooxidation reactions (Fig. 4a and Appendix A Fig. S3), the higher mass concentration of the limonene + NO₂ photooxidation (Fig. 4b and Appendix A Fig. S3) might be attributed to the unique chemical structure and reaction characteristics of limonene, as well

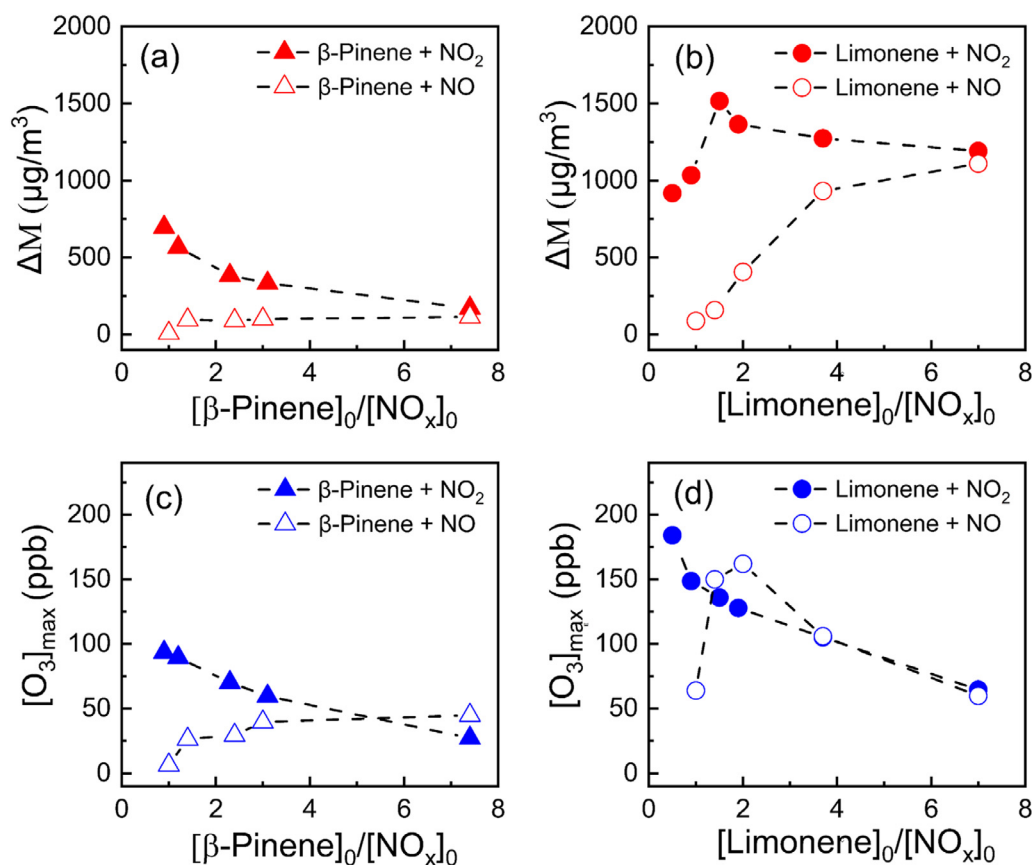


Fig. 5 – The particle mass concentration (ΔM) and O_3 maximum concentration ($[O_3]_{\max}$) as a function of $[\beta\text{-pinene}]_0/[NO]_0$, $[\beta\text{-pinene}]_0/[NO_2]_0$, $[\text{limonene}]_0/[NO]_0$, and $[\text{limonene}]_0/[NO_2]_0$.

as the faster rates of O_3 formation from NO_2 photolysis (Fig. 4c and d).

While previous studies have examined the impact of the $[\beta\text{-pinene}]_0/[NO_x]_0$ and $[\text{limonene}]_0/[NO_x]_0$ ratios on the SOA formation (Liu and Shi, 2021), the specific effects of the $[\beta\text{-pinene}]_0/[NO]_0$, $[\beta\text{-pinene}]_0/[NO_2]_0$, $[\text{limonene}]_0/[NO]_0$, and $[\text{limonene}]_0/[NO_2]_0$ ratios remain unexplored. Hence, the dependence of particle mass concentration, O_3 maximum concentration, and SOA yields on the $[\beta\text{-pinene}]_0/[NO]_0$, $[\beta\text{-pinene}]_0/[NO_2]_0$, $[\text{limonene}]_0/[NO]_0$, and $[\text{limonene}]_0/[NO_2]_0$ has been studied and the results are illustrated in Fig. 5 and Appendix A Fig. S4, respectively. For the β -pinene + NO_x photooxidation reactions, as the $[\beta\text{-pinene}]_0/[NO]_0$ ratio increases, the particle mass concentration increases slightly (Fig. 5a), which trend holds true for the O_3 accumulation (Fig. 5c). In contrast, the particle mass concentration and O_3 maximum concentration decrease with the increase of the $[\beta\text{-pinene}]_0/[NO_2]_0$ ratio (Fig. 5a and c). For the limonene + NO photooxidation reactions, the particle mass concentration remarkably increases with increasing the $[\text{limonene}]_0/[NO]_0$ ratio from 0.9 to 3.7 and slightly increases with further increasing the $[\text{limonene}]_0/[NO]_0$ ratio to 7.0 (Fig. 5b), revealing a more pronounced suppression effect of NO on the limonene photooxidation than that on the β -pinene photooxidation. Note that the reaction of NO with the RO_2 radical could form a stable $RONO_2$ product, which consumes the RO_2 radical and thereby inhibits the autooxidation process.

The aforementioned different effects of the $[\beta\text{-pinene}]_0/[NO]_0$ and $[\text{limonene}]_0/[NO]_0$ ratios on the SOA formation could be rationalized that the reaction of NO with the RO_2 radicals generated from limonene is more readily than that with the RO_2 radicals generated from β -pinene. The effect of $[\text{limonene}]_0/[NO_2]_0$ on the cumulative O_3 concentration (Fig. 5d) is also more pronounced than $[\beta\text{-pinene}]_0/[NO_2]_0$ (Fig. 5c). Such difference in the response of O_3 formation from β -pinene and limonene provides a reference for studies on the nonlinear relationship O_3 formation to various precursors in real atmospheric conditions (Liu and Shi, 2021; Sillman, 1999).

2.2. VUV-FEL photoionization mass spectra of β -pinene and limonene photooxidation

In this study, chemical compositions of the compounds formed from the β -pinene + NO/NO_2 and limonene + NO/NO_2 photooxidation processes were measured using a VUV-FEL photoionization aerosol mass spectrometer. Previous studies have shown that by optimizing the VUV-FEL wavelengths and pulse energies, the threshold photoionization can be achieved with high sensitivity, high selectivity, and negligible fragmentation (Normile, 2017; Zang et al., 2022; Zhang et al., 2024, 2025). Our prior studies indicated that reaction times for terpenes and NO_x at slightly higher concentrations can be remarkably reduced without significant variation of chemical compositions of SOA (Zang et al., 2022; Zhang et al., 2024,

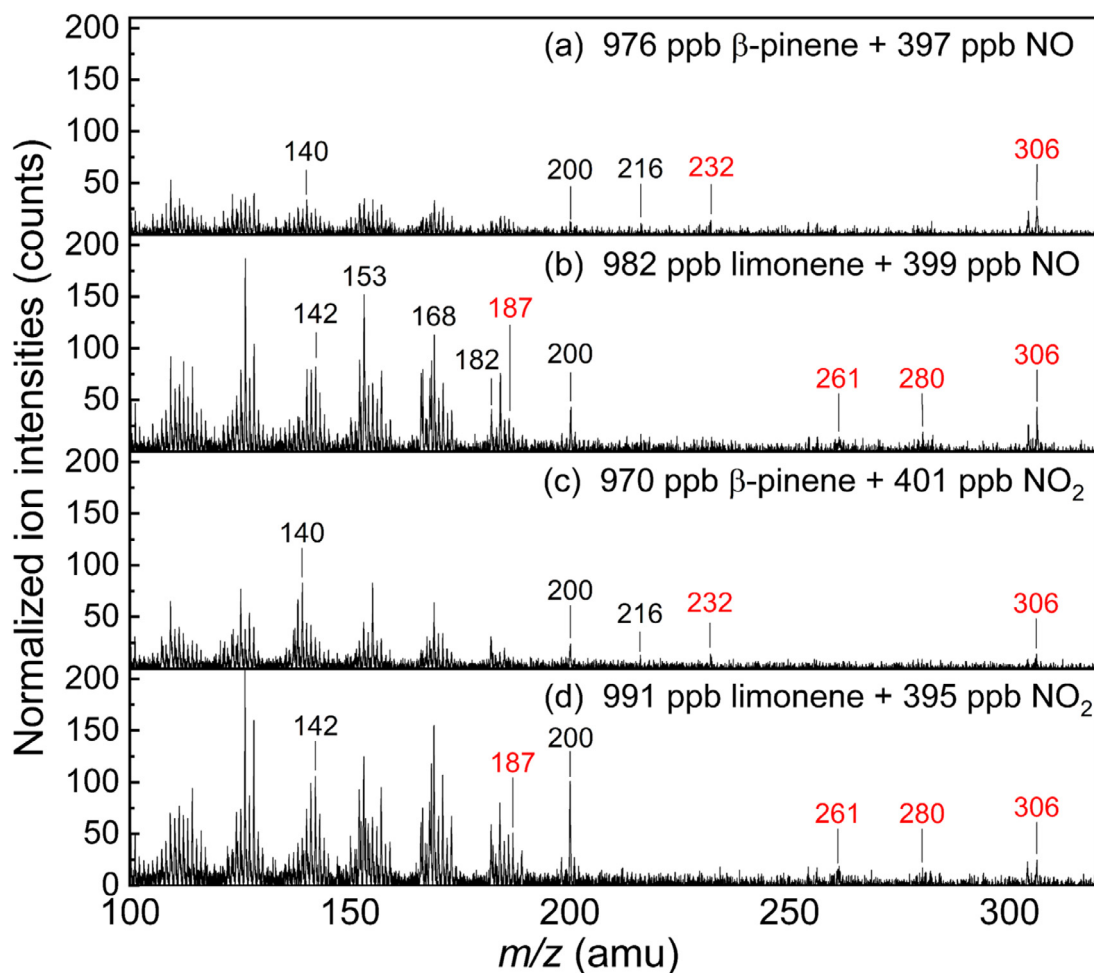


Fig. 6 – VUV-FEL photoionization mass spectra of the compounds generated under different experimental conditions. The compounds were ionized by the VUV-FEL at 125.0 nm. The newly-observed peaks are marked in red.

2025). In order to promote the beamtime efficiency of the expensive VUV-FEL, we prefer to employ VUV-FEL photoionization mass spectrometry to analyze these compounds under slightly higher concentrations than natural atmospheres.

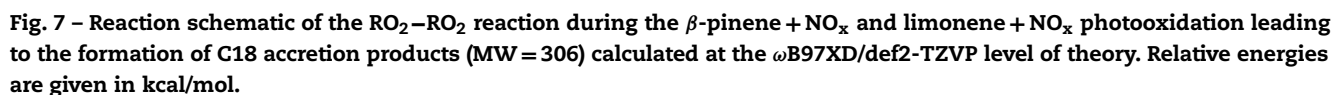
2.3. Characterization of chemical compositions during β -pinene and limonene photooxidation

In this work, the VUV-FEL aerosol mass spectra were recorded every 5 min. Time point of maximum particle mass concentration was select to show the mass spectra in Fig. 6 as the SOA products. The number and intensities of mass spectral peaks of the compounds generated from limonene + NO/NO₂ photooxidation (Fig. 6b and d) are found to be significantly higher than those from β -pinene + NO/NO₂ (Fig. 6a and c), which complexity trend of chemical compositions supports the trend of particle mass concentrations and particle number concentrations (Figs. 1–4). The compounds observed at MW = 140, 168, 182, and 200 are consistent with the major products of β -pinene and limonene oxidation as previously determined by Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR MS) and electrospray ionization mass spectrometry (ESI-MS) analysis (Huang et al., 2023;

Liu et al., 2022). A new compound at MW = 306 is observed in all the photooxidation processes of β -pinene + NO/NO₂ and limonene + NO/NO₂. While a new compound at MW = 232 is observed in the β -pinene + NO/NO₂ photooxidation (Fig. 6a and c), three new compounds at MW = 187, 261, and 280 are observed in the limonene + NO/NO₂ photooxidation (Fig. 6b and d). The peak positions in the β -pinene + NO photooxidation (Fig. 6a) are similar to those of β -pinene + NO₂ photooxidation (Fig. 6c), with slightly higher intensities in the β -pinene + NO₂ photooxidation. A similar trend is observed for limonene + NO/NO₂ (Fig. 6b and d). This indicates that the relative abundance of NO and NO₂ in the atmosphere would affect the mass concentrations of SOA, but does not significantly alter the diversity of chemical compositions.

2.4. Characterization of chemical compositions during β -pinene and limonene photooxidation

In general, OH and O₃ (generated by NO_x photolysis) are the main oxidants for the VOC + NO_x photooxidation. Since the OH concentration is difficult to be precisely estimated under the current experimental conditions and O₃ has been demonstrated to play an important role in the VOC + NO_x photooxi-



Recent studies have underscored the significance of accretion reactions in the generation of low-volatility dimer com-

pounds that serve as fundamental components of SOA formation (Berndt et al., 2018; Kenseth et al., 2023; Perakyla et al., 2023; Shi et al., 2022). However, limited information is available to the structural characteristics and formation mechanisms of accretion products, particularly those derived from different monoterpenes. In the atmospheric context, the RO₂ radicals engage in the accretion reactions, specifically manifested as RO₂ + R'O₂ → ROOR' + O₂ and R(CO)O₂ + R'O₂ → R(CO)OR' + O₂, leading to the formation of peroxide and ester accretion products. Fig. 7 shows the mechanisms for the generation of newly-observed products MW = 306 initiated from the ozonolysis of β-pinene and limonene. The terminal alkene functionality of β-pinene undergoes a reaction with O₃ to yield the primary ozonide (POZ) (B1-1), which is exothermic with a predicted value of 51.1 kcal/mol at the ωB97XD/def2-TZVP level of theory. B1-1 undergoes decomposition reaction (with the release of HCHO) to form B1-2, which is exothermic by 16.9 kcal/mol

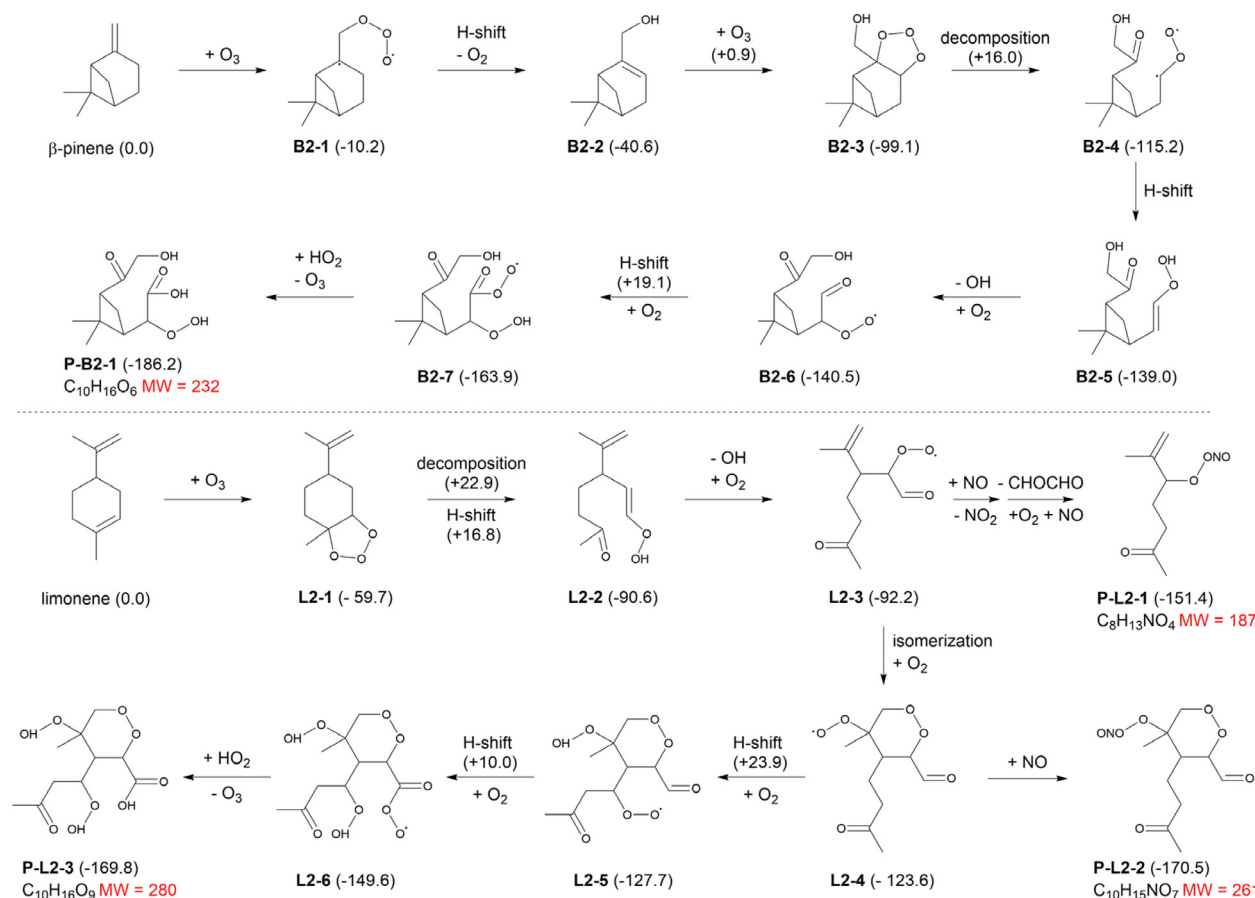


Fig. 8 – Autooxidation pathways of the β -pinene + NO_x and limonene + NO_x reactions calculated at the ω B97XD/def2-TZVP level of theory. Relative energies are given in kcal/mol.

and overcomes a barrier of 13.9 kcal/mol. B1-2 undergoes intramolecular H-shift to produce B1-3 with a predicted exothermicity of 13.7 kcal/mol and barrier of 15.5 kcal/mol. B1-3 releases OH and then reacts with O₂ to form B1-4 with a predicted exothermicity of 6.1 kcal/mol. The B1-4 → P-B1-1 (MW = 306) reaction is highly exothermic by 38.2 kcal/mol. Similar to the β -pinene → B1-4 process, the limonene → L1-4 process is also thermodynamically exothermic and dynamically feasible. In particular, the L1-4 → P-L1-1 (MW = 306) reaction for the limonene system (exothermic by 98.2 kcal/mol) is much more favorable than the corresponding reaction for the β -pinene system (exothermic by 38.2 kcal/mol). Previous research has shown that ester accretion products are generally less reactive and more stable than peroxide accretion products (Kenseth et al., 2023; Perakyla et al., 2023). Consequently, even though β -pinene and limonene undergo photooxidation to form compounds of the same molecular weight, limonene tends to yield ester dimeric accretion products, which is beneficial to the SOA formation with higher particle mass concentrations than β -pinene (Figs. 1–4). Establishing the chemical basis for formation of ester and peroxide accretion products helps to link the atmospheric degradation of different monoterpenes to the formation of low-volatility byproducts that can drive aerosol formation and growth.

Recent research has revealed that the monoterpene can rapidly produce large amounts of low-volatility vapors

through gaseous autooxidation processes (Bianchi et al., 2019). With this clue, the possible formation routes for O₃ addition to the double bonds of β -pinene and limonene were explored by considering the potential reactions in the autooxidation pathway and the main results are shown in Fig. 8. The β -pinene → B2-2 process is exothermic with a predicted value of 40.6 kcal/mol, which aligns with previous results (Iinuma et al., 2007). Similarly, the B2-2 → B2-3 process is highly exothermic by 58.5 kcal/mol with an extremely low barrier of 0.9 kcal/mol. B2-3 is decomposed to form B2-4 with a predicted exothermicity of 16.1 kcal/mol and barrier of 16.0 kcal/mol. Similar to the B1-2 → B1-4 process, the B2-4 → B2-6 process is also exothermic with a predicted value of 25.3 kcal/mol. B2-6 undergoes intramolecular H-shift to form B2-7 with a predicted exothermicity of 23.4 kcal/mol and barrier of 19.1 kcal/mol. The B2-7 → P-B2-1 (MW = 232) reaction is exothermic by 22.3 kcal/mol. The limonene → L2-3 is substantially exothermic by 92.2 kcal/mol. The process from L2-3 to their corresponding product organic nitrates P-L2-1 (MW = 187) is facile with a calculated exothermicity of 59.2 kcal/mol. Due to the presence of the intramolecular double bond, L2-3 undergoes isomerization to form L2-4 with a predicted exothermicity of 31.4 kcal/mol. The L2-4 → P-L2-2 (MW = 261) reaction is feasible with a calculated exothermicity of 46.9 kcal/mol. L2-4 undergoes intramolecular H-shift and then reacts with O₂ to form L2-5 with a calculated

exothermicity of 4.1 kcal/mol and barrier of 23.9 kcal/mol. The L2-5 $\rightarrow$ L2-6 process is also feasible with a calculated exothermicity of 21.9 kcal/mol and barrier of 10.0 kcal/mol. The L2-6 $\rightarrow$ P-L2-3 (MW = 280) reaction is a favorable process with a calculated exothermicity of 20.2 kcal/mol.

3. Conclusion and implication

In this study, we investigated the individual effects of initial NO and NO₂ concentrations on the size distribution and particle mass concentration of SOA formed from the photooxidation of isomeric monoterpene, β -pinene and limonene. We further employed the VUV-FEL photoionization aerosol mass spectrometry to analyze the distinct chemical components of SOA. As compared to β -pinene, the SOA formed from the limonene photooxidation exhibits a higher particle number concentration and mass concentration and is more significantly influenced by the initial NO_x levels. For the β -pinene + NO_x photooxidation, a monotonous suppression effect of [NO]₀ on the particle mass concentration and a monotonous enhancement effect of [NO₂]₀ on the particle mass concentration is observed, respectively, which show significant correlation with the variation of cumulative amount of O₃ generated by NO_x photolysis. In contrast, for the limonene + NO_x photooxidation, a monotonous suppression effect of [NO]₀ on the particle mass concentration and a parabolic suppression/enhancement effect of [NO₂]₀ on the particle mass concentration is observed, respectively; as [NO]₀ gradually increases, the cumulative O₃ produced by NO_x photolysis exhibits a nonlinear trend, and there is no apparent correlation between the particle mass concentration and the accumulated O₃ concentration. Such difference of O₃ accumulation in the photooxidation reactions of different monoterpenes is consistent with the nonlinear variation of ground-level O₃ concentrations on NO_x levels (Chelani, 2010; Liu and Shi, 2021). Our laboratory studies confirm that NO_x influences the photochemical reactions leading to the formation of O₃ and SOA, substantiating its pivotal role in atmospheric oxidation capacity.

Using the advantages of high photoionization efficiency and wide wavelength modulation range of VUV-FEL, the photoionization mass spectra of SOA produced by the β -pinene and limonene photooxidation were detected to analyze their chemical compositions. A series of new compounds at MW = 232 and 306 formed from the β -pinene + NO_x photooxidation and those at MW = 187, 261, 280, and 306 formed from the limonene + NO_x photooxidation are identified, respectively. The MW = 306 (P-B1-1 and P-L1-1) compounds observed in the β -pinene + NO_x and limonene + NO_x photooxidation are generated by O₃ addition to the double bond and is assigned to peroxide and ester accretion product, respectively. The difference in the particle mass concentration and SOA yield for the β -pinene and limonene photooxidation may be ascribed to the diversity of product types formed through different oxidation pathways. The MW = 232 (P-B2-1), 187 (P-L2-1), 261 (P-L2-2) and 280 (P-L2-3) products are generated separately through autooxidation pathways of β -pinene and limonene. The observation of organic nitrates, MW = 261 (P-L2-2), resulting from limonene photooxidation, reflects the

NO-inhibited effect on autooxidation pathway as observed in the experiments. This work underscores the significance of O₃ addition on the double bonds of β -pinene and limonene in the NO_x photooxidation process. Our findings reveal significant differences in the accretion products and HOMs formation derived from the photooxidation of β -pinene and limonene. The necessity to incorporate NO_x concentration-driven processes in the understanding of SOA formation and its dynamic evolution is a pressing concern in the context of future climate models, particularly in the neighborhood of emission origins.

Declaration of competing interest

The authors have no conflicts to disclose.

CRediT authorship contribution statement

Yingqi Zhao: Writing – original draft, Software, Methodology, Investigation, Formal analysis, Data curation. **Zhaoyan Zhang:** Investigation, Formal analysis, Data curation. **Ya Zhao:** Investigation, Formal analysis, Data curation. **Chong Wang:** Visualization, Investigation, Formal analysis, Data curation. **Hua Xie:** Validation, Methodology, Funding acquisition, Formal analysis. **Jiayue Yang:** Resources, Methodology. **Weiying Zhang:** Resources, Methodology. **Guorong Wu:** Resources, Methodology. **Gang Li:** Writing – review & editing, Writing – original draft, Validation, Supervision, Project administration, Investigation, Funding acquisition, Formal analysis, Data curation, Conceptualization. **Ling Jiang:** Writing – review & editing, Writing – original draft, Visualization, Validation, Supervision, Project administration, Investigation, Funding acquisition, Formal analysis, Data curation, Conceptualization. **Xueming Yang:** Validation, Supervision, Project administration, Funding acquisition.

Acknowledgments

The authors gratefully acknowledge the Dalian Coherent Light Source (DCLS) and Specreation Co., Ltd. for support and assistance. This work was supported by the [National Natural Science Foundation of China](#) (Nos. 22125303, 92361302, 92061203, 22103082, 22273101, 22288201, and 21327901), the National Key Research and Development Program of China (No. 2021YFA1400501), the Innovation Program for Quantum Science and Technology (No. 2021ZD0303304), Dalian Institute of Chemical Physics ([DICP I202437](#)), Chinese Academy of Sciences (No. [GJJSTD20220001](#)), and the International Partnership Program of CAS ([121421KYSB20170012](#)).

Appendix A Supplementary data

Supplementary material associated with this article can be found in the online version at [doi:10.1016/j.jes.2024.08.007](https://doi.org/10.1016/j.jes.2024.08.007).

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